

Synthesis of Some New Thiazolidone- Derivatives with Possible Fungicidal Activities

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Abstract: The present study was conducted to evaluate the antimicrobial activity, many phenols and compounds with phenolic groups have antifungal potency. A large number of fungicides are formulated as wettable powders; this is the form most commonly used for spray mixes. Modern wettable powders are easily wetted and disperse well in water. They simply inhibit fungus growth temporarily. If the fungus is freed from such substance, it would revive. Such a chemical is called a "fungistat" and the phenomenon of temporarily inhibiting the growth is "fungistasis". Some other chemicals, like certain phenanthrene derivatives and Bordeaux mixture, may inhibit spore production without affecting the growth of vegetative fungistate hyphae. These are called "antisporeulants". 2-amino 4-Phenyl thiazole is condensed with appropriate ethanol and piperidine aromatic was refluxed on water bath for 1 hr. Various benzylidines and thiazolidones respectively and synthesized compounds showed moderate to good antifungal activity with respect to standard drugs.

Keywords: 2-amino 4-Phenyl thiazole, EtOH, anhydrous zinc chloride, antifungal activity

INTRODUCTION

Schiff bases appear to be an important intermediate in a number of enzymatic reactions involving interaction of an enzyme with an amino or a carbonyl group of the substrate. One of the most important types of catalytic mechanism is the biochemical process which involves the condensation of a primary amine in an enzyme usually that of a lysine residue, with a carbonyl group of the substrate to form an imine, or Schiff base. Stereochemical investigation carried out with the aid of molecular model showed that Schiff base formed between methylglyoxal and the amino group of the lysine side chains of proteins can bend back in such a way towards the N atom of peptide groups that a charge transfer can occur between these groups and oxygen atoms of the Schiff bases. Heterocyclic chemistry is currently experiencing a renaissance because of the interest in heterocyclic chemistry is currently experiencing a renaissance because of the interest in heterocyclic scaffolds as templates for combinatorial chemistry. They are known to possess a variety of biological activities such as analgesic, anti-inflammatory, protein kinase C inhibitor.⁴ Many pyrazole

derivatives possess remarkable antiepileptic and antimicrobial,⁵ antiamebic,⁶ Azetidinones, commonly known as beta-lactams, are well-known heterocyclic compounds among the organic and medicinal chemists. The activity of the famous antibiotics such as penicillin, cephalosporin, monobactams and carbapenems are attributed to the presence of azetidinone ring in them. Azetidinone can be prepared from Schiff's bases, which are the condensation products of aldehydes and amino compounds. They are considered significant owing to their wide range of biological application. Recently, some other types of biological activity besides the antibacterial activity have been reported in compounds containing azetidinone ring. Such biological activities include antimicrobial. The structures of the various synthesized compounds were assigned on the basis of IR, ¹H-NMR spectral Nitrogen containing heterocyclic with sulfur atom is an important class of compounds in medicinal chemistry. Thiazoles being an integral part of many potent biologically active molecules such as sulfathiazole (Antimicrobial drug), Ritonavir (Antiretroviral drug), Abafungin (Antifungal drug) with trade

name Abase cream and Bleomycin and Tiazofurin (Antineoplastic drugs) have been explored previously

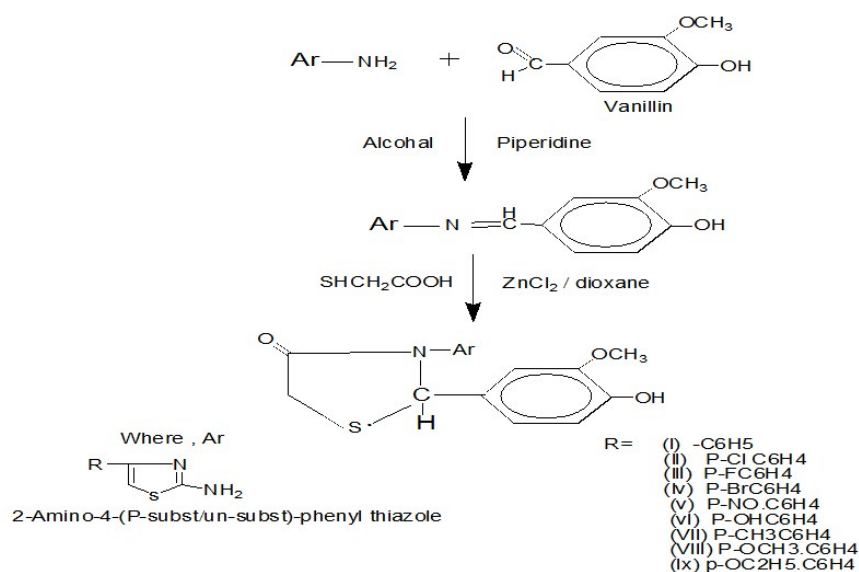
MATERIAL AND METHODS

Thiazoles are important class of natural and synthetic compounds. Thiazole derivatives display a wide range of biological activities such as cardiotoxic, fungicidal, sedative, anesthetic, bactericidal and anti-inflammatory. The synthesis of thiazole derivatives is important of their wide range of pharmaceutical and biological properties. A large number of fungicides are formulated as wettable powders; this is the form most commonly used for spray mixes. Modern wettable powders are easily wetted and disperse well in water. A wetting agent is usually present in most wettable powder formulations, but the adding of a spreader-sticker is sometimes desirable, especially on plants with glossy or waxy leaves. Fungicide which is effective only if applied prior to fungal infection is called a protectant; e.g. zaneb, sulphur, etc. On the other hand, fungicide which is capable of eradicating a fungus after it has caused infection and thereby "curing" the plant, is called a therapeutant. 8-quinolinol, antibiotics like Aureofungin, etc. Eradicants are those which remove pathogenic fungi from an infection court. Some chemicals do not kill fungi. The IR spectra were recorded on IR affinity-1, DRS-8000A, Shimadzu, Ptc. Ltd., Japan spectrophotometer. The $^1\text{H-NMR}$ was recorded in DMSO on Bruker Advance II 400 MHz spectrometer using TMS as an internal standard. Melting points were determined in open capillary tubes and are

uncorrected. The purity of the compounds was checked by TLC-using Silica gel-G (Merck). Column chromatography was performed on silica gel. All the compounds were tested for their antibacterial and antifungal activities by broth dilution method. Nitrogen containing heterocyclic compounds has received considerable attention due to their wide range of pharmacological activity. The pyrazoles and the pyrimidines constitute interesting class of organic compounds with diverse chemical and biological application. They are known to possess variety of biological activities such as analgesic, anti-inflammatory, protein kinase inhibitor. Many pyrazole derivatives possess remarkable antiepileptic and antimicrobial, antiamoebic,

SYNTHESIS OF 2-AMINO THIAZOLE

A solution of 76 gm of thiourea in 200 ml warm water is placed in a 500 cc and three necked flask equipped with dropping funnel, mechanical stirrer and reflux condenser. 143 gm (122 cc) of alpha beta dichloroethyl ether is added and mixture is heated under gentle reflux with stirring for 2 hours as the reaction proceeds. The two layers gradually merge to the cold solution, sufficient solid NaOH is added to liberate 2-Amino thiazole from its salt. Ether is added to dissolve the product and ether is evaporated. 2-Amino thiazole is recrystallised from ethanol. Yield (80%), M.P. 91°C, % of N and S Calculated 16.27% and 37.20%, % of N and S found 16.20% and 36.20%. The compound has been characterized on the basis of I.R spectra and the purity of the compound was checked by T.L.C.



[1] SYNTHESIS OF [L] N-(4-PHENYL-2-THIAZOLYL)-2-IMINO-(4'-HYDROXY-3'-METHOXY BENZYLIDINE):

A mixture of 2-Amino-4-phenyl thiazole (0.01 mol) and vanillin 0.01 moles in ethanol 30 ml and piperidine 3-4 drops was refluxed on water bath for 1 hours. the reaction mixture was cooled and the solid separated was filtered and recrystallised from ethanol. Yield: (92%), m.p. 1600C, IR(KBr) = 1210-1220 cm⁻¹ (due to C-O-c), 1665-1670 cm⁻¹, (C=N), 1590 - 1595 cm⁻¹ (C=C), 3000-3110 cm⁻¹ (due to -OH), 1640-1625 cm⁻¹ and 1250 cm⁻¹ (due to C=N and C-N), PMR = δ 4.0-4.2(3H,s,OCH₃) , δ 7.1-7.6(8H, m, ArH) , δ 8.2-8.5(1H,s =CH) , δ 9.5-9.7(1H, s,-OH) Similarly , various N-[4-(P-subst/un-subst)-Phenyl-2-thiazolyl]-2-imino-(4'-Hydroxy-3'-methoxy benzylidene were prepared by using similar reaction procedure and their analytical data are incorporated in the table(1) respectively.

[2] SYNTHESIS OF N-[4-PHENYL-2-

THIAZOLYL]-2'-[4-HYDROXY-3-METHOXYPHENYL]-4"-THIAZOLIDONE

To a mixture of compound first (0.01 moles) and mercaptoacetic acid 0.001 mole) dissolved in dioxane (20ml) a pinch of anhydrous zinc chloride was added and the mixture was refluxed for eight hours on cooling the separated solid was washed with dilute sodium bicarbonate was crystallized from ethanol .Yield 47 , M.P 1480c IR(KBr) = 3100 cm⁻¹ (due to OH) , 1640-1625 cm⁻¹ AND 1250, (C=N), 1590 - 1595 cm⁻¹ (C=C), 3000-3110 cm⁻¹ (due to -OH), 1640-1625 cm⁻¹ and 1250 cm⁻¹ (due to C=N and C-N), 1210-1220 CM-1(due to C-O-C), 1660-1670 cm⁻¹ (due to C-S-C) 1685 cm⁻¹ (due to cyclic > c=O), PMR = δ 3.82-3.86(3H, s,OCH₃), δ 9.85(1H, s,OHm), δ 4.15-(2H, s CH₂S), δ 6.5-6.8(1H, s,-CH), δ 6.5-7.4(8H, M,-Ar-H), Similarly, various N-[4-(P-subst/un-subst)-Phenyl-2-thiazolyl]-2-imino-(4'-Hydroxy-3'-methoxyphenyl)-4" thiazlidone were prepared by using similar reaction procedure and their analytical data are incorporated in the table(II) respectively.

Table- 1 Analytical data of N-[4-(P-subst/un-subst)-Phenyl-2-thiazolyl]-2-imino-(4'-Hydroxy-3'-methoxy benzylidene.

S.N.	Nature of Ar	Molecular Formula	Yield %	M.P. OC	ELEMENTAL ANALYSIS			
					% of N		% of S	
					Cald	Fond	Cald	Found
la	2-Amino-4-phenyl thiazole	C ₁₇ H ₁₄ N ₂ O ₂ S	42	138	09.03	09.00	10.32	10.25
lb	2-Amino-4(p-chloro)-phenyl thiazole	C ₁₈ H ₁₁ N ₂ O ₂ ClS	50	140	19.92	19.86	22.77	22.69
lc	2-Amino-4(p-fluoro)-phenyl thiazole	C ₁₇ H ₁₃ N ₂ O ₂ SF	52	145	08.53	08.50	09.75	09.70
ld	2-Amino-4(p-bromo)-phenyl thiazole	C ₁₇ H ₁₃ N ₂ O ₂ SBr	48	106	07.21	07.11	08.24	08.20
le	2-Amino-4(p-nitro)-phenyl thiazole	C ₁₈ H ₁₁ N ₃ O ₄ S	47	148	11.83	11.76	09.01	08.93
lf	2-Amino-4(p-hydroxy)-phenyl thiazole	C ₁₇ H ₁₄ N ₂ O ₃ S	48	165	08.53	08.49	09.75	09.73
lg	2-Amino-4(p-methyl)-phenyl thiazole	C ₁₉ H ₁₄ N ₂ O ₂ S	52	226	08.64	08.60	09.87	09.80
lh	2-Amino-4(p-methoxy)-phenyl thiazole	C ₁₉ H ₁₄ N ₂ O ₂ S	53	246	08.23	08.20	09.41	09.35
li	2-Amino-4(p-ethoxy)-phenyl thiazole	C ₁₉ H ₁₄ N ₂ O ₂ S	50	250	07.90	07.80	09.03	09.00

Table - 2 Analytical data of N-[4-Phenyl-2-thiazolyl]-2'-[4-hydroxy-3-methoxyphenyl]-4"-thiazolidone

S.N.	Nature of Ar	Molecular Formula	Yield %	M.P. OC	ELEMENTAL ANALYSIS			
					% of N		% of S	
					Cald	Fond	Cald	Found
lla	2-Amino-4-phenyl thiazole	C ₂₀ H ₁₄ N ₂ O ₃ S ₂	52	179	07.09	07.05	16.24	16.22
llb	2-Amino-4(p-chloro)-phenyl thiazole	C ₁₈ H ₁₃ N ₂ O ₃ S ₂ Cl	53	190	06.50	06.45	14.93	14.90
llc	2-Amino-4(p-fluoro)-phenyl thiazole	C ₂₀ H ₁₃ N ₂ O ₃ S ₂ F	50	185	06.54	06.50	15.05	15.00
lld	2-Amino-4(p-bromo)-phenyl thiazole	C ₂₀ H ₁₃ N ₂ O ₃ S ₂ Br	42	186	05.93	05.88	13.55	13.52
lle	2-Amino-4(p-nitro)-phenyl thiazole	C ₂₀ H ₁₃ N ₄ O ₅ S ₂	50	225	09.56	09.49	14.97	14.55
llf	2-Amino-4(p-hydroxy)-phenyl thiazole	C ₂₀ H ₁₄ N ₂ O ₄ S ₂	42	148	06.82	06.78	15.60	15.55
llg	2-Amino-4(p-methyl)-phenyl thiazole	C ₂₁ H ₁₆ N ₂ O ₃ S ₂	51	144	06.86	06.81	15.68	15.60
llh	2-Amino-4(p-methoxy)-phenyl thiazole	C ₂₁ H ₁₆ N ₂ O ₄ S ₂	50	141	06.60	06.55	15.09	14.55
lli	2-Amino-4(p-ethoxy)-phenyl thiazole	C ₂₂ H ₁₈ N ₂ O ₄ S ₂	54	145	06.39	06.20	14.61	14.02

ANTIFUNGAL SCREENING

The newly synthesized compounds were evaluated against *Alternaria alternata* fungus at optimum

temperature of 28± 1oC (after 7 days incubation) was observed. After inoculation, All the petridishes were incubated at (25 ± 20C) for 7 days, the

efficiency of various anti-fungal was recorded by measuring the radial growth of the fungal colony (in mm). The percentage inhibition of fungus mycelia growth was calculated by the equation.

$$\% \text{ of Inhibition} = \frac{[(C - T) \times 100]}{C}$$

Where C and T are average colony diameters (in mm) of the fungal colony in control (C) and treated (T) plates respectively. and their Antifungal screening data are incorporated in the table(III) respectively.

Table-3 Effect of Some Newly Synthesised Antifungal Compounds against *Alternaria alternata* at optimum temperature (After 7 days incubation)

Compound	Dose	Average colony diameter (in mm) in PDA medium	% Inhibition
Control		60.88	
la	0.20	2.8	95.39
lb	0.20	3.2	94.73
lc	0.20	4.0	93.42
ld	0.20	1.9	96.87
le	0.20	2.7	95.55
lf	0.20	2.8	95.39
lg	0.20	9.9	83.71
lh	0.20	3.0	95.06
li	0.20	3.2	94.73
lla	0.20	3.1	94.90
llb	0.20	2.7	95.55
llc	0.20	4.1	93.25
lld	0.20	3.5	94.24
lle	0.20	3.2	94.73
llf	0.20	2.8	95.39
llg	0.20	2.4	96.05
llh	0.20	1.7	97.20
lli	0.20	2.8	95.39
BAVISTIN(Std drug)	0.20	0.22	99.65

RESULT AND: DISCUSSION

It is evident from fungal screening data that all the newly synthesized compound tested were found satisfactorially superior over control but inferior to that of standard antifungal (Bavistin) compound. Mostly synthesized compound showed marked of the fungal growth in vitro test. It can also be concluded from the result that mostly synthesized compound are good antifungal and showed significant level of antifungal activity and compound No(lg) showed moderate activity.

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REFERENCES

1. A. L. Barry, "The antimicrobial susceptibility Test Principles and Practices, Edited by Illus lea and Fabiger 180, Bio. Abstr. 64 (1976). 25183.
2. Al-Saddi. M. S; Faidallah. H. M; Rostom. S. A. F. Arch. Pharm. Che. Life Sci., 2008,
3. Anandarajagopal, K., Sunilson, A. J., Illavarasu, A., Thangavelpandian, N., Kalirajan, R.,. Int. J. ChemTech. Res., 2010, 2(1), 45.
4. Abid, M., Azam, A., Bioorg. Med. Chem. Lett., 2006, 16(10), 2812
5. B. Das and M. k. Rout. J Indian .Chem. Soc. 1955. 32. 668.
6. Baselt. T; Rehse. K. Archived der pharmazie., 2008, 24, 645-654. 1-Ashok, K., Rathod., IDiseant. J. PharmTech. Res. 2011, 3(1), 435.
7. Doherty, A. M., Ann. Rep. Med. Chem. 2004, 39, 335. 4 Mariappan, G., Saha, B. P.,. Sutharson, L., Haldar, A., Ind. J Chem., 2010, 49B, 1671.
8. Eichenberger, K., Schweizer, E., Schimdt, P., US Patent.
9. Heckmann, G., Bach, T., Angew. Chem. Int. Ed. 2005, 44, 1199 6 Millan, S. E., . Nicholas, D. P., Bioorg. Med. Chem.

- Lett., 2009, 19 222 .
- Hansford, K. A., Zanzarova, V., Dorr, A., Lubell, W. D., J. Comb. Chem. 2004, 6, 893.
 - Krishna, R., Pande, B. R., Bharthwal, S. P., Parmar, S. S., Eur. J. Med. Chem., 1980, 15, 567.
 - L.F. Firser, " Experiments in Organic Chemistry " D.C. Health and Company Boston, . 1957, 3rd Edn. p 180.
 - Regaila, H. A., El-Bayonk, A. K., Hammad, M., Egypt J. Chem., 1979, 20, 197.
 - Sharma H.L., Principle of Pharmacology, 2, 748-752 (2008).
 - Schimdt, P., Eichenberger, K., Schweizer, E., German Offen., 196908479, Chem. Abstr. 1970, 72, 31837u .
 - Tripathi. K. D. Essential of medical pharmacology., 2003, 5th edition, 627-686.
 - Karpov K. A; Nazarenko A.V; Pekarevskii B.V; potekhin. V. M. Russian journal of applied chemistry. 2001, 74, 998-1001.
 - Marchant R. Jaysukhal and D.S Chothia . J. Med. Chem, 1970 . 13.335: M.S. Shinagare and D.B. Ingle. J. Indian chem. Soc. 1976. 53. 1036..
 - Ulusoy. N; Kiraz. M; Kucukbasmaci. O. Monatshefte fur chemie., 2002, 133, 1305-1315.
 - Kaplancikli. Z. A; Zitouni. G. T; Revial. G; Guven. K. Arch pharm Res., 2004, 27, 10811085.
 - Lester A., Itschep M. and Foye, Principle of Medicinal Chemistry, 5, 833 (2002)
 - Y.L. Nene, Fungicidal Plant diseases Control Oxford and I.B.H. Publishing Company, New Delhi, 1971.
 - Amr, Ael-G., Abdel-Lalif, N. A., Abdalla, M. M., Bioorg. Med. Chem., 2006, 14(2), 373.



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